Tetrahedron Letters, Vol.26, No.48, pp 5931-5934, 1985 0040-4039/85 \$3.00 + .00 Printed in Great Britain ©1985 Pergamon Press Ltd.

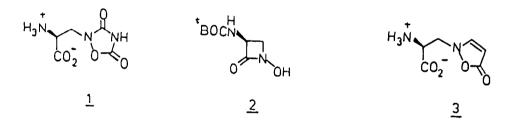
SYNTHESIS OF L- $\beta$ -(ISOXAZOLIN-5-ONE-2-YL)-ALANINE: A NOVEL METHOD FOR THE SYNTHESIS OF N-SUBSTITUTED 3,4-UNSUBSTITUTED ISOXAZOLIN-5-ONES'

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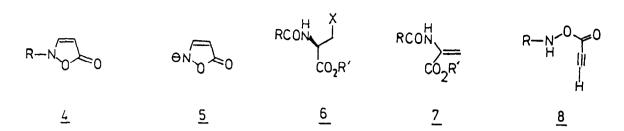
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<u>Abstract</u>: A mild new general method for the synthesis of the isoxazolin-5-one-2-yl ring system (<u>4</u>) via a favoured 5-endo-dig cyclisation is described; this methodology has been successfully applied for the first chemical enantio-efficient synthesis of  $L-\beta-(isoxazolin-5-one-2-yl)-alanine (<u>3</u>).$ 

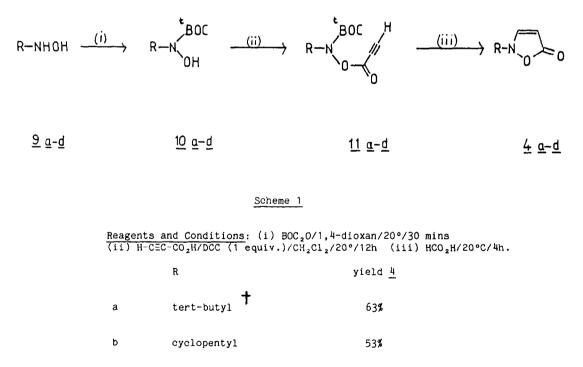
We have recently described a general method for the enantiospecific synthesis of  $\beta$ -amino alanine derivatives,<sup>2</sup> as exemplified by a total synthesis of the neuroexcitatory amino acid quisqualic acid (1) from the known  $\beta$ -lactam 2. We now wish to report the first synthesis of L- $\beta$ -(isoxazolin-5-one-2-yl)-alanine (3), a highly base labile amino acid isolated from the roots of pea seedlings, <sup>3</sup>,<sup>4</sup> from the same  $\beta$ -lactam precursor 2 and describe a novel method for the construction of the heterocyclic moiety of this amino acid.



There are few reports concerning the synthesis of <u>N</u>-substituted, 3,4- unsubstituted isoxazolin-5-ones (<u>4</u>). Most satisfactory is the alkylation of the anion <u>5</u><sup>5</sup>, which gave modest yields (13-50%) of <u>4</u> when alkylated with good alkylating agents such as dimethylsulphate (R=Me). However the authors attempts to synthesise <u>3</u> via displacement on a  $\beta$ -substituted alanine (<u>6</u>) failed, presumably due to the facility with which such substrates undergo  $\beta$ -elimination to give dehydroalanines (<u>7</u>). Reaction of <u>N</u>-substituted hydroxylamines with propiolic esters or  $\beta$ -ketoesters by refluxing in pyridine has led to 3- or 4-substituted isoxazolin-5-ones,<sup>6</sup> but no report has been made for the 3,4-unsubstituted case. Any potential synthesis of <u>3</u> must also be mild, due to the lability of the heterocyclic ring towards dilute alkali or strong acid.<sup>3</sup>,<sup>6</sup>



A new approach to this system was therefore sought. It was considered that <u>N</u>-substituted, <u>O</u>-propiolated hydroxylamines (<u>8</u>) could furnish isoxazolin-5-ones (<u>4</u>) in a 5-endodig manner.<sup>7</sup> To this end, a series of <u>N</u>-substituted hydroxylamines (<u>9 a-d</u>) were treated with di-tert-butylpyrocarbonate (BOC<sub>2</sub>O) prior to acylation on oxygen with propiolic acid and dicyclohexylcarbodiimide (DCC), to produce <u>11 a-d</u> (~70% following chromatography). Treatment of crude or chromatographed <u>11 a-d</u> with neat formic acid afforded the isoxazolin-5-ones (<u>4</u>) directly in 53-66% overall yield from the hydroxylamines <u>9 a-d</u> (Scheme 1) (Table 1). All gave characteristic<sup>5</sup> <sup>1</sup>H NMR spectra (5 and 8 p.p.m., d, J 3.5Hz) and showed strong absorption in both the UV ( $\lambda_{max} \sim 265$  nm,  $\varepsilon \sim 10^{\circ}$ ) and IR (CHCl<sub>3</sub>) ( $\nu_{c=0}$  1730 cm<sup>-1</sup>) spectrum.



c methyl 66%

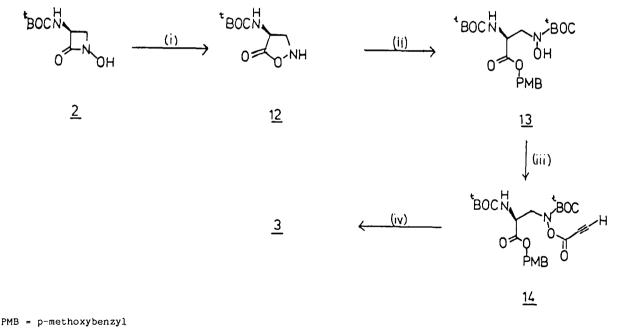
d benzyl 57%

Table 1

The utility of this methodology has been exemplified by the synthesis of the amino acid 3. Thus the  $\beta$ -lactam 2° was isomerised to the known isoxazolidin-5-one 12°,° by treatment with catalytic sodium ethanethiolate (83%). 12 was converted to 13 in one pot by treatment with aqueous caesium carbonate<sup>10</sup>, followed by addition of BOC<sub>2</sub>O and esterification (after freeze-drying) with p-methoxybenzyl chloride. 13 was obtained in 53% yield from 12 following chromatography. Acylation of 13 with propiolic acid and DCC gave the <u>O</u>-propiolyl-hydroxylamine 14 (77%) which underwent deprotection and spontaneous cyclisation on exposure to neat formic acid to give the amino acid 3. Purification by H.P.L.C. (reverse phase octadecasilane, using 1% formic acid in water as eluant) afforded pure 3 (61% from 14),  $[\alpha]^2\beta$  -57.5° (C 0.4 in H<sub>2</sub>O) [Lit.,<sup>11</sup> -62.7° (C 1.4 in H<sub>2</sub>O)];  $\nu_{max}$  (KBr disc) 3500-2500 mbr, 1700s, 1630-1590s, and 1530s cm<sup>-1</sup> (Lit.<sup>3</sup> identical);  $\delta_{\rm H}$  (500 MHz, D<sub>2</sub>O) 3.98 (1H, t, J 5Hz), 4.19 (2H, d, J 5Hz), 5.15 (1H, d, J 3.5Hz) and 8.11 p.p.m. (1H, d, J 3.5Hz);  $\delta_{\rm C}$  (125 MHz, D<sub>2</sub>O) 55.0 (t), 55.8 (d), 90.3 (d), 157.1 (d), 173.3 (s) and 177.2 (s) p.p.m.; m/z (positive argon fast atom bombardment) 173 (M<u>H</u><sup>+</sup>, base peak). The synthetic sample thus obtained was identical by <sup>1</sup>H NMR,  $\lambda_{max}$  and m.p. (and mixed m.p.) to an authentic sample of the natural material.<sup>11</sup>

Synthesised without recourse to initial protection on nitrogen, <u>viz N</u>-t-butylhydroxylamine/ H-CEC-CO<sub>2</sub>H/DCC/CH<sub>2</sub>Cl<sub>2</sub>/20°C/12 hours, then usual DCC work up.

+ NMR yield prior to H.P.L.C. ca 81%.



Scheme 2

<u>Reagents and Conditions</u>: (i)  $EtS^{\Theta}Na^{\oplus}$  (2.5 mol %, THF, 20°C, 72h); (ii) (a)  $Cs_2CO_3$  [1 mol equivalent, THF/H<sub>2</sub>O (1:1), 20°C, 4h] (b)  $BOC_2O$  (1eq, 1 h) then freeze-dry (c) PMB-Cl [1·1 equiv., NaI(cat), DMF, 40°C, 24 h]; (iii) HCEC-CO<sub>2</sub>H (1 equiv.) DCC (1 equiv., CH<sub>2</sub>Cl<sub>2</sub>, 20°C, 12 h); (iv) HCO<sub>2</sub>H (neat, 20°C, 8 h) then water/ether partition and freeze-drying.

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(Received in USA 27 June 1985)